



The Jizan Carbon Mineralization Pilot: Lessons Learned from Lab to Field

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NIM Industry Insights Series - Beyond Capture: CO₂
Mineralisation and Carbon Recycling for a Net-Zero Industry

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King Abdullah University of
Science and Technology

جامعة الملك عبدالله
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Outline

- Background

Jizan Pilot Deployment :

- Field Assessment
- Lab Investigation and Modeling
- Pilot Design & Execution
- Key Results

Inorganic Geological Carbon Cycle



Carbonate-silicate cycle

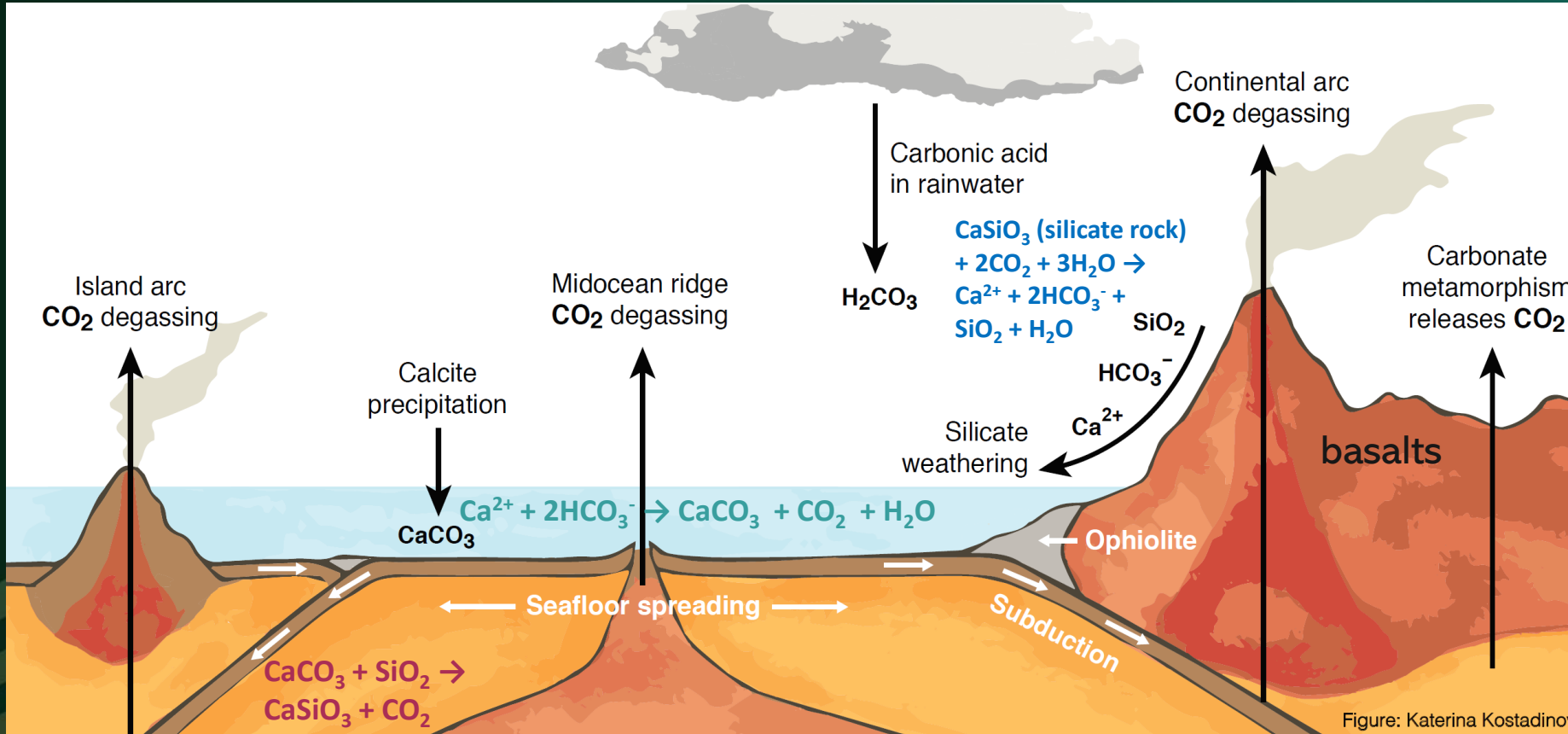
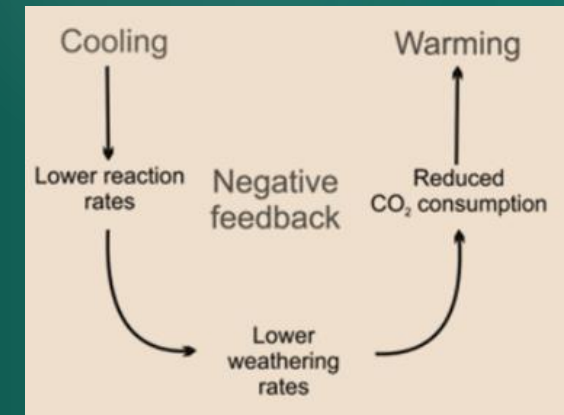
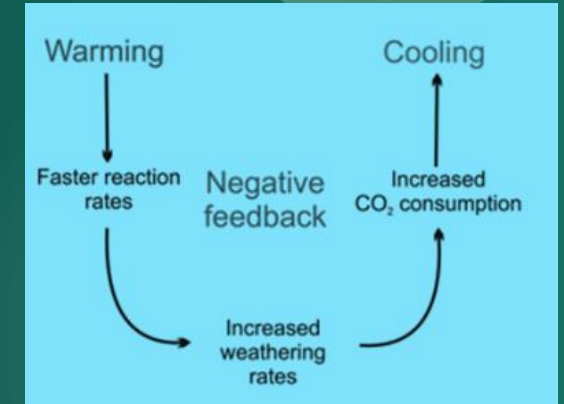


Figure: Katerina Kostadinova

Kasting, 2019

Earth's Thermostat



Walker, Hays, Kasting, 1981

https://cce.nasa.gov/ocean_biology_biogeochemistry/ocean_worlds/about.html

CO₂ Mineralization in Carbonate Minerals



CARBON DIOXIDE DISPOSAL IN CARBONATE MINERALS[†]

KLAUS S. LACKNER^{†,§} & CHRISTOPHER H. WENDT[¶]
DARRYL P. BUTT[‡], EDWARD L. JOYCE, JR.[‡] & DAVID H. SHARP[‡]

[‡]LOS ALAMOS NATIONAL LABORATORY 1995

Proposed “a safe and permanent method for CO₂ disposal based on combining CO₂ chemically with abundant raw materials for stable carbonate minerals (calcium and magnesium carbonates)”

Two approaches to the carbonation of calcium and magnesium oxides :

- 1- Direct carbonation at high temperature
- 2- Carbonation in the aqueous phase (low temperature)

Rock	MgO, wt%	CaO, wt%	R _C	R _{CO₂}	
Peridotites	Dunite	49.5	0.3	6.8	1.8
	Harzburgite	45.4	0.7	7.3	2.0
	Lherzolite	28.1	7.3	10.1	2.7
Serpentinite	~40	~0	~8.4	~2.3	
Gabbro	~10	~13	~17	~4.7	
Basalt	Continental tholeiite	6.2	9.4	26	7.1

Lackner et al. 1995

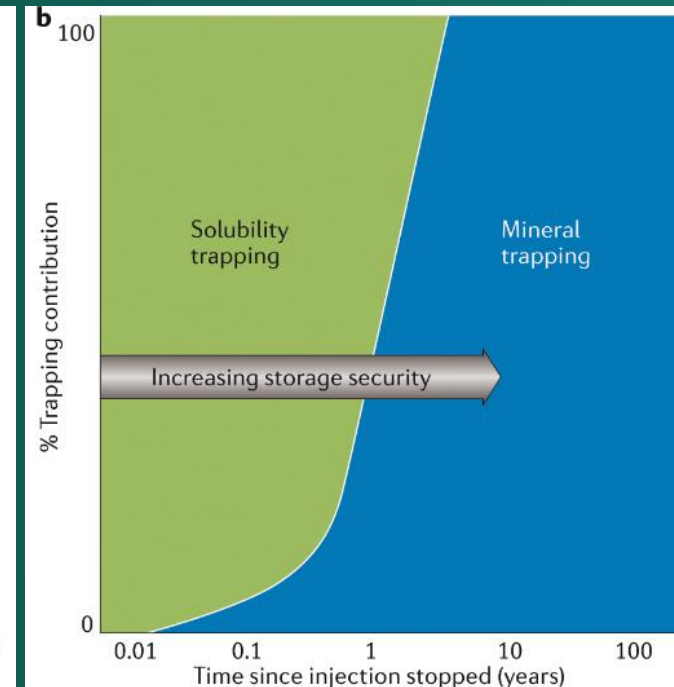
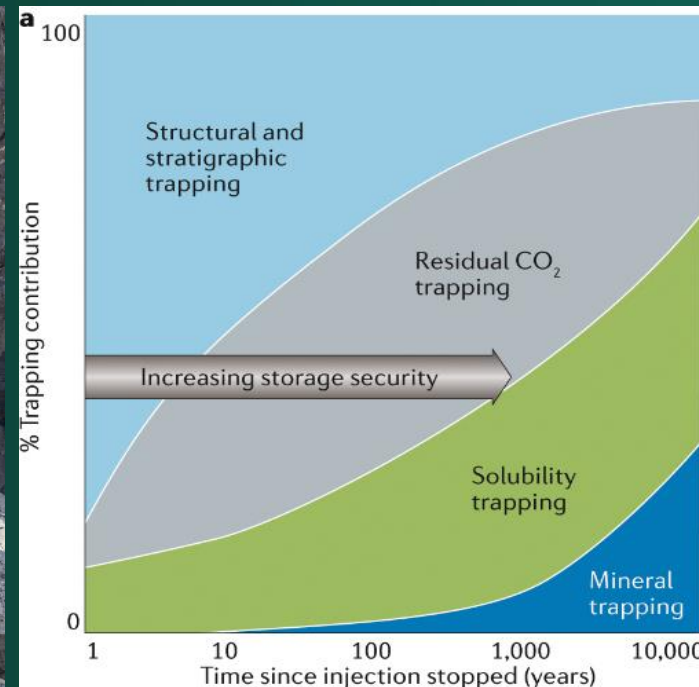
CO₂ Mineralization in Basalts



- Dark-colored, fine-grained, igneous rock
- Porous, but generally has **poor permeability**
- Mainly composed of **plagioclase** and **pyroxene**
- **Rich in divalent cations** such as Ca²⁺ and Mg²⁺ that react with CO₂ to form stable carbonate minerals e.g. calcite, magnesite

Pilots (>100 tons):

- ✓ Wallula Washington, USA (2013)
- ✓ Carbfix, Iceland (2012)
- ✓ Jizan, KSA (2023-2024)



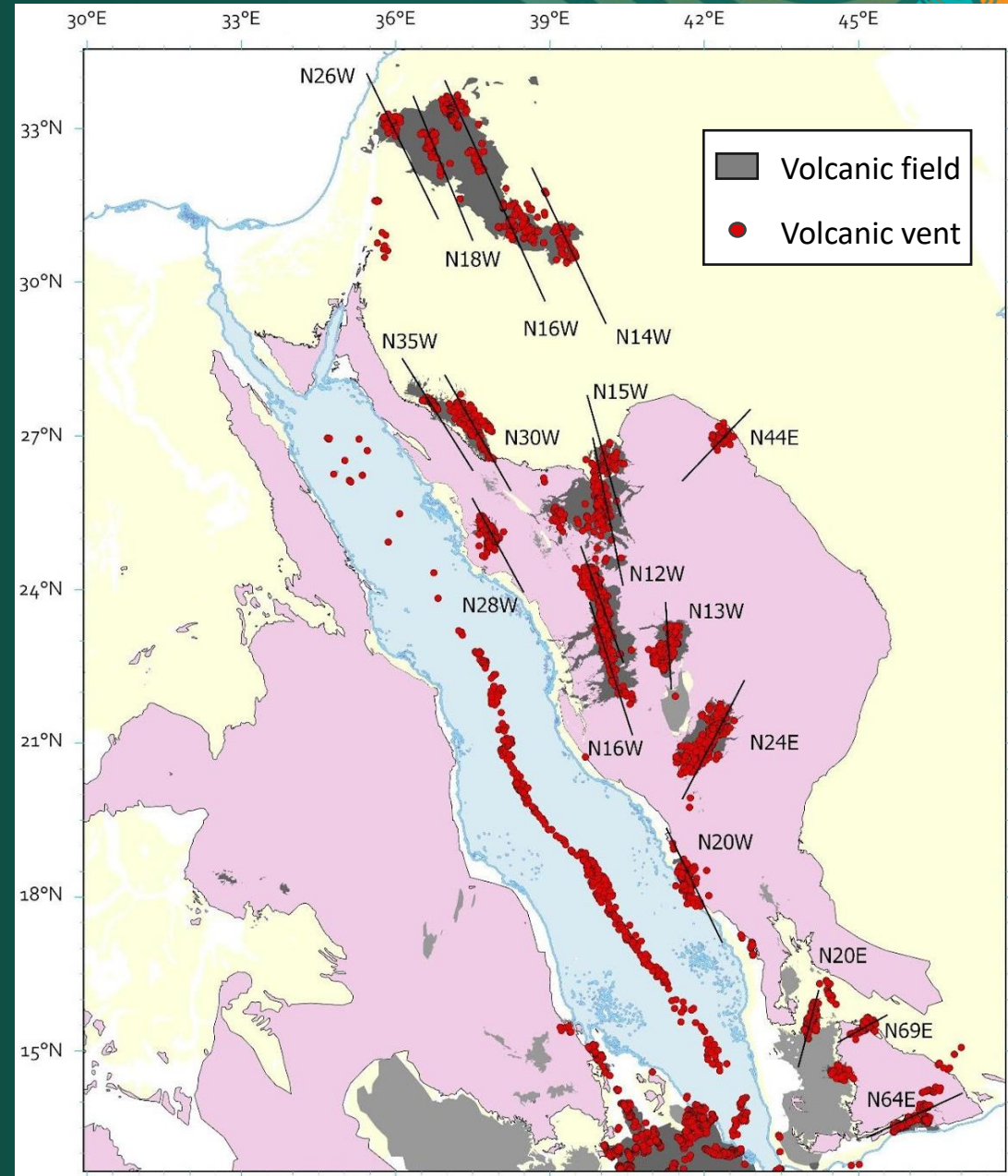
(Snæbjörnsdóttir et al., 2020)

Basalts in Saudi Arabia

- Saudi Arabia is rich in volcanic (Harrat) and mafic formation in the western region
- Basalt and other reactive formations have the potential to remove CO₂ permanently



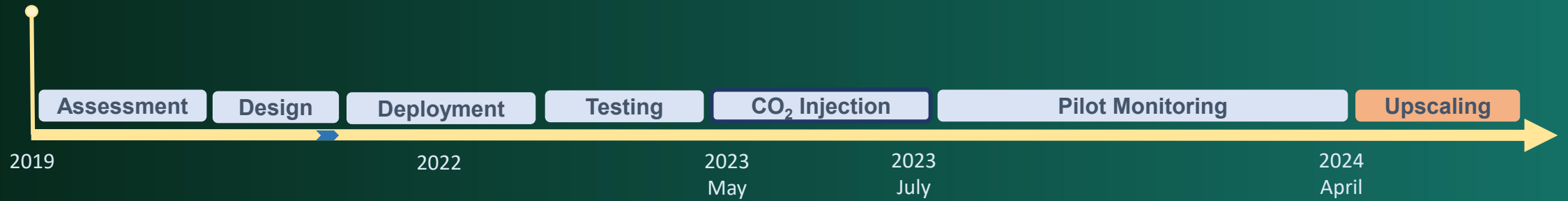
Jizan Group Basalts



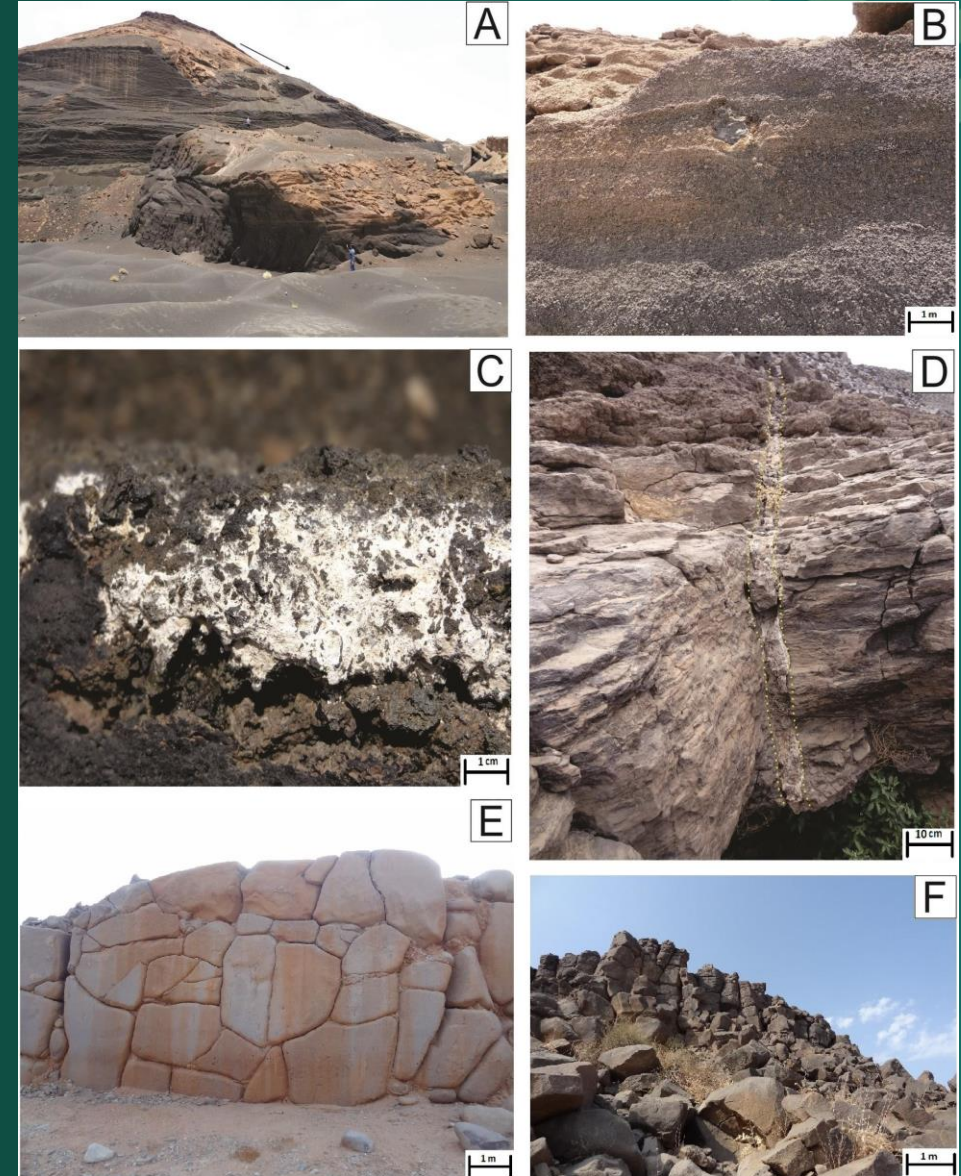
Jizan CO₂ Mineralization Pilot



Initial Field Work



Field Assessment (Resources)



Basalt Characterization (suitability)

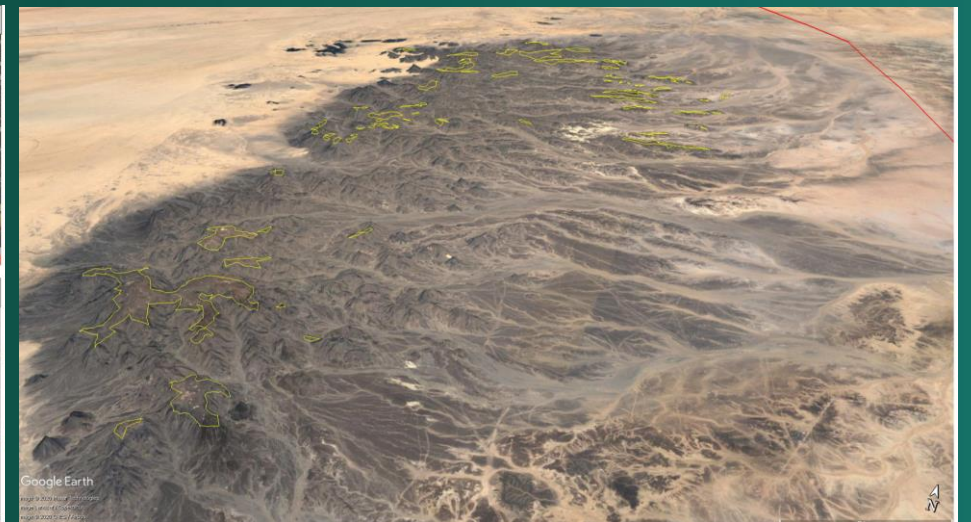
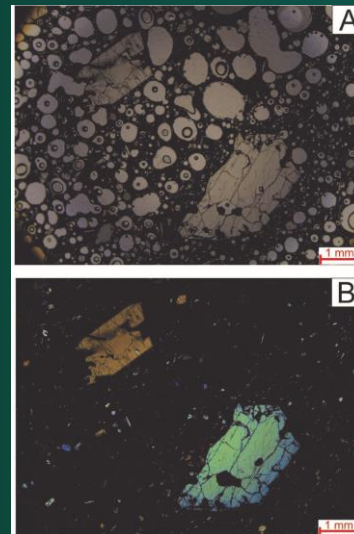


Petrova & Afifi, 2020



Name of the rock	Volume % in rock	Plagioclase composition % anorthite	Volume fraction of anorthite	Weight mineral kg/m^3	Weight fraction Ca in mineral	Ca in mineral kg/m^3	Kilomoles Ca/m^3 rock
31A altered basalt plagioclase	15	30	0.045	123.75	0.25	31.17	0.78
33A fresh porous unaltered basalt plagioclase	10	62	0.062	170.50	0.25	42.95	1.07
Name of the rock	Volume % in rock	Pyroxene composition % diopside	Volume fraction of diopside	Wt mineral in kg/m^3	Weight fraction Ca in mineral	Ca in mineral kg/m^3	Kilomoles Ca/m^3 rock
31A altered basalt clinopyroxene	5	85	0.043	144.50	0.19	26.75	0.95

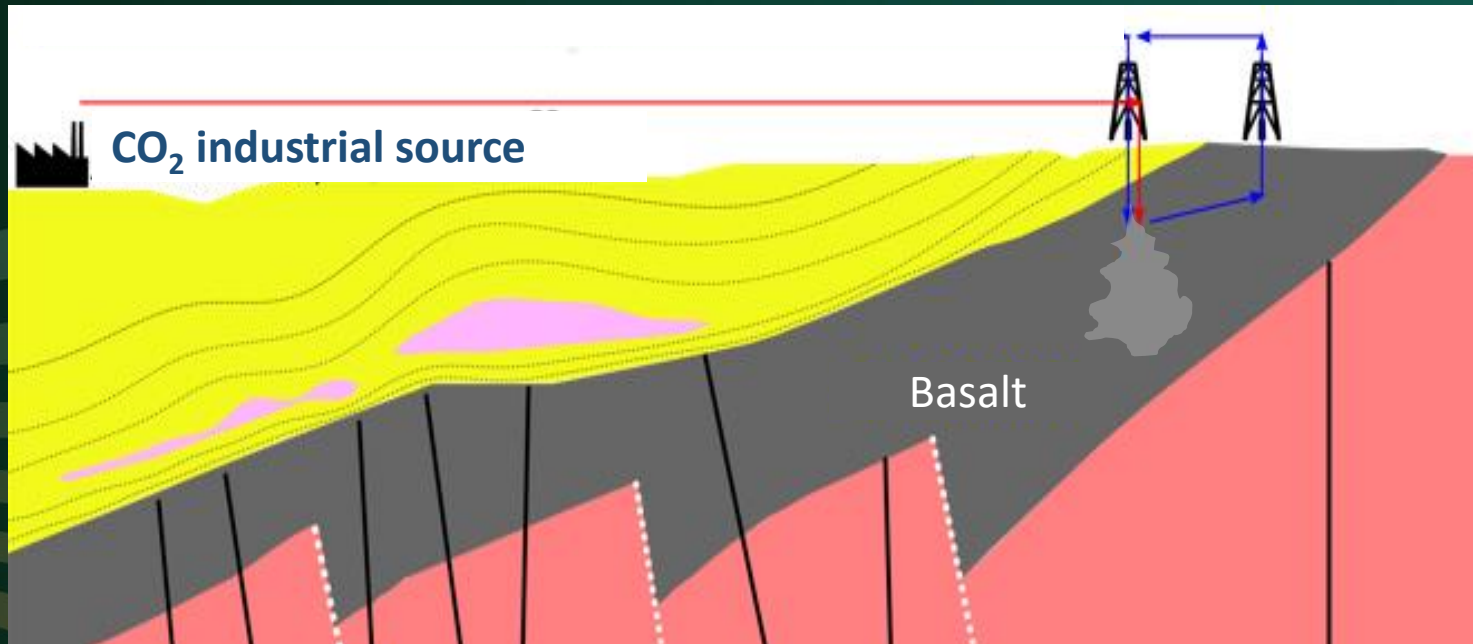
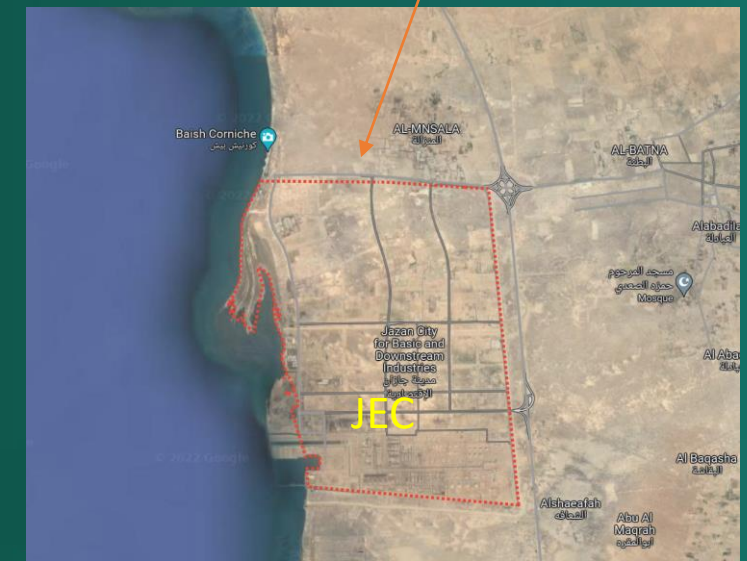
Table 5.1: Calcium content in plagioclase and pyroxene from Harrat Rahat



Pilot Selection in Jizan

The Jizan basalt is well suited for carbon mineralization :

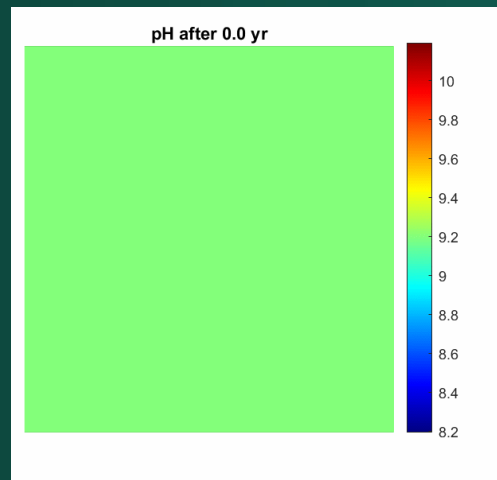
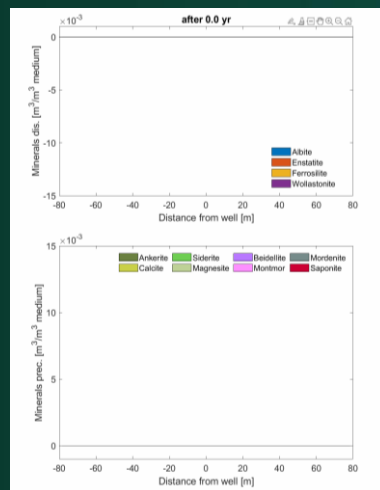
- Geologic setting is favorable and well known
- Contain large volume of basalt at various depths
- Close to sources of CO₂ and SO₂ in Jizan Economic City
- Seawater is available for injecting CO₂ into the basalt reservoir



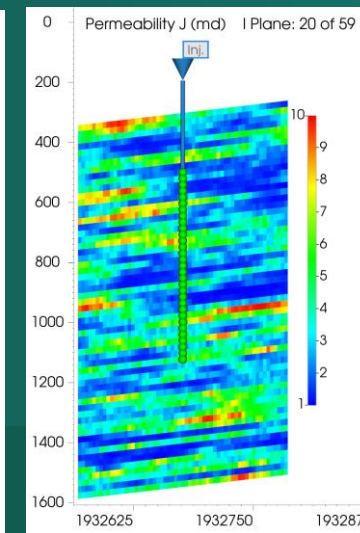
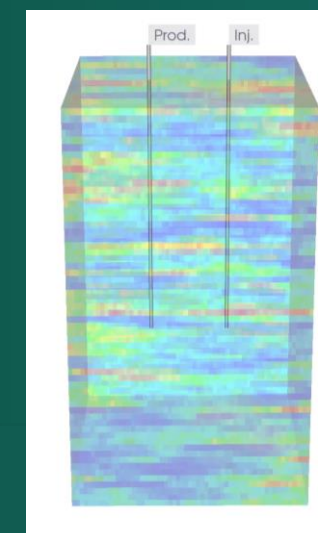
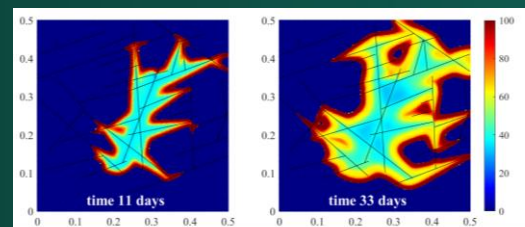
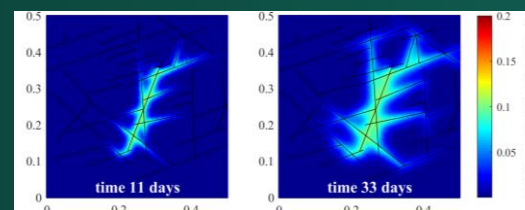
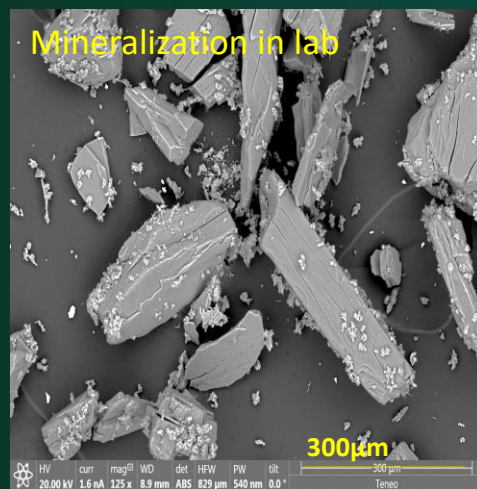
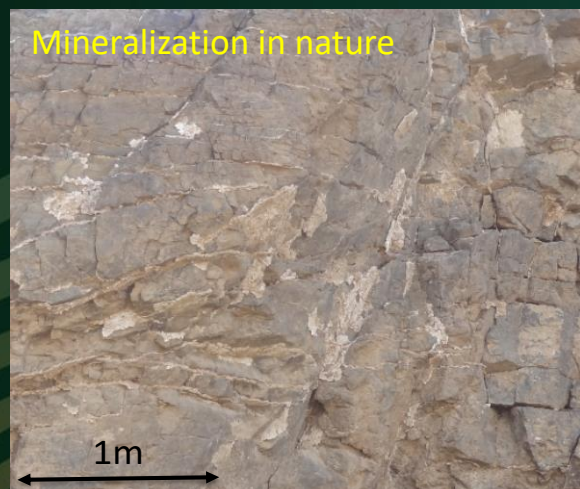
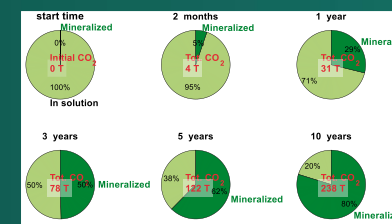
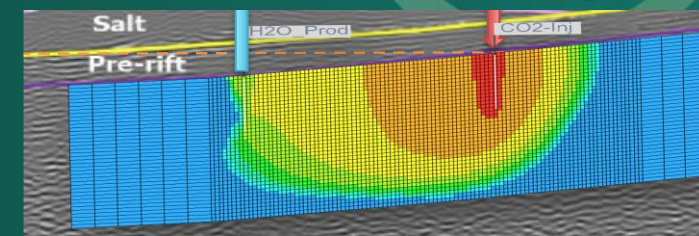
Pilot Design: Lab and Modeling Studies



Lab-scale modeling



Field-scale modeling

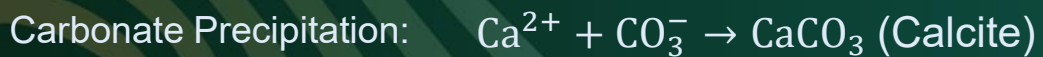
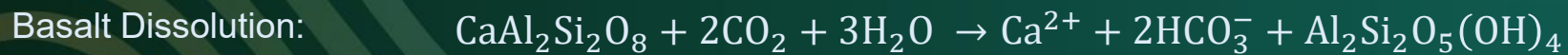
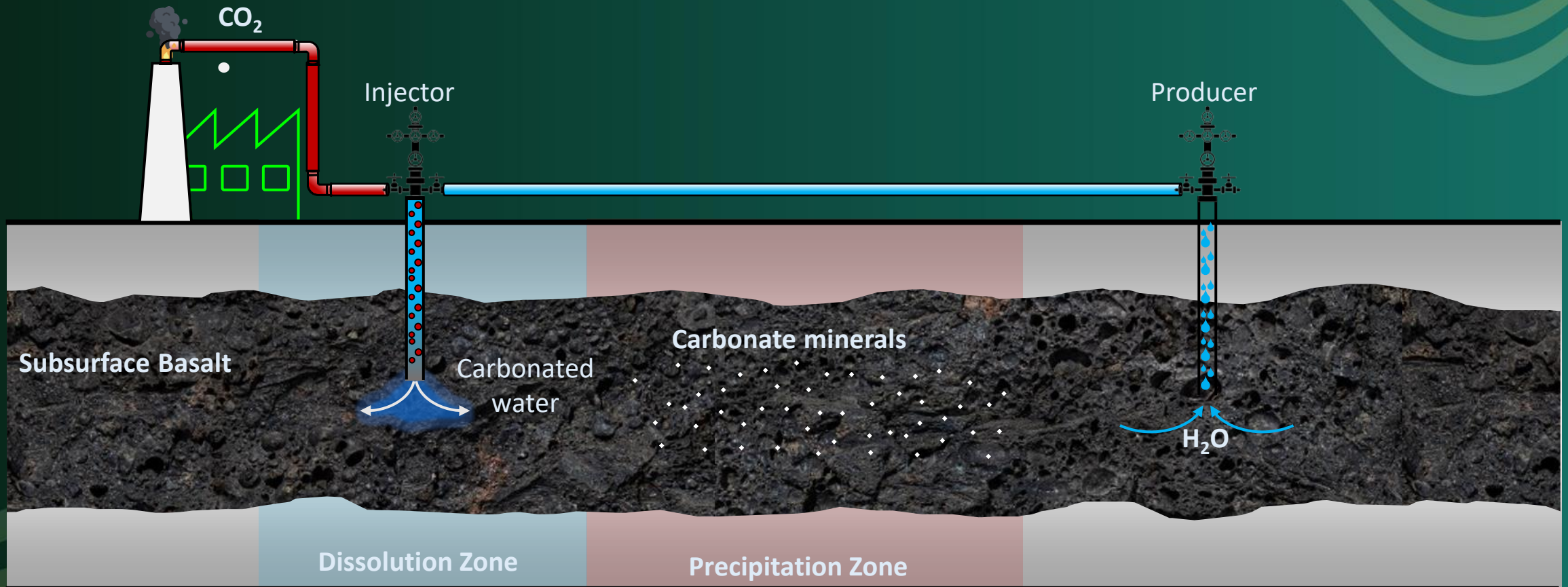


Omar, Addassi, Berno, ..., Oelkers, Hoteit (2025). An experimental study of the mineral carbonation potential of the Jizan Group basalts. *J of Greenhouse Gas Control*

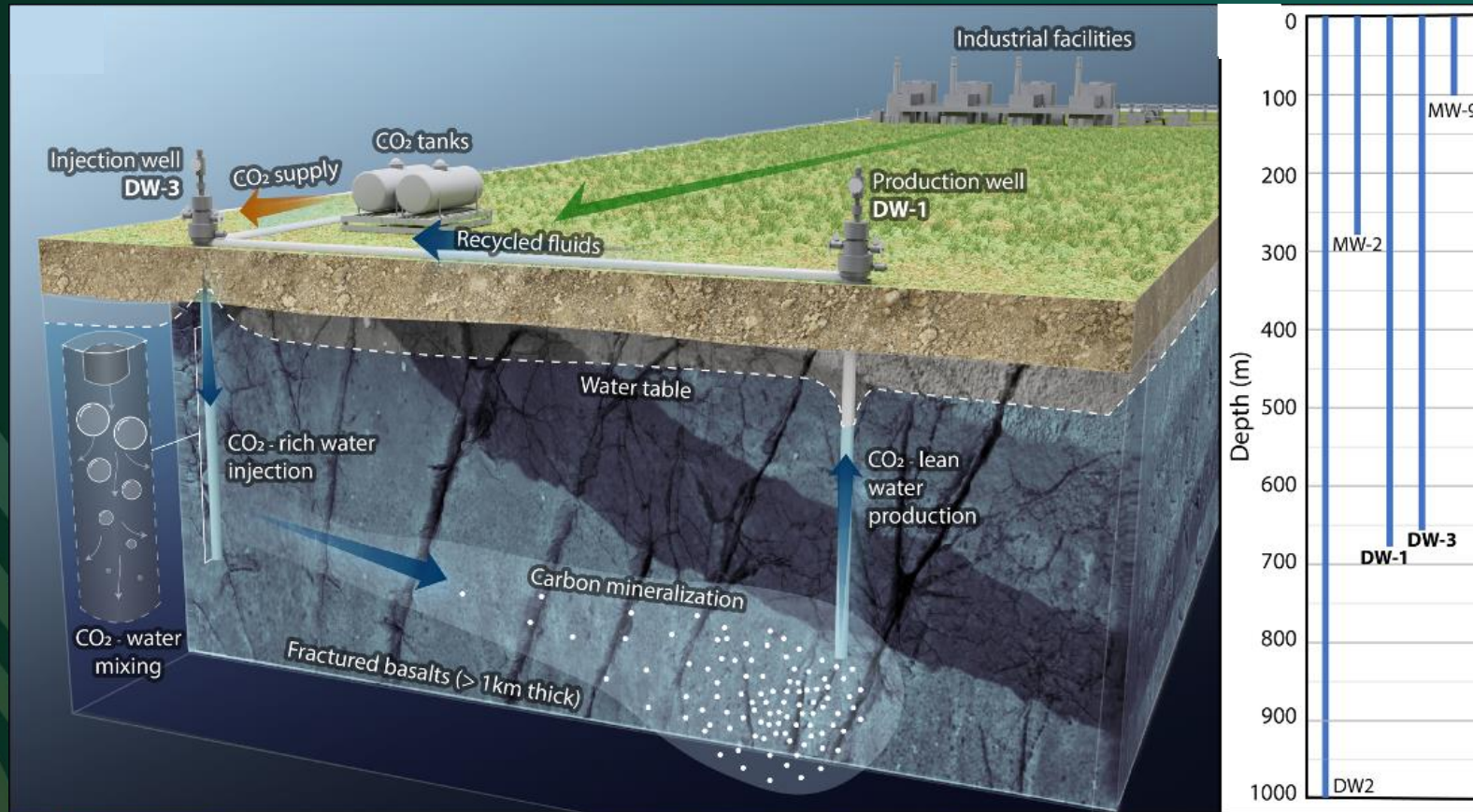
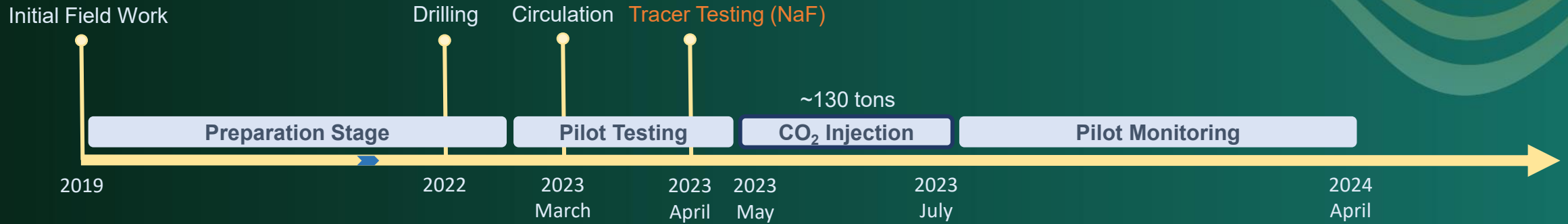
Alqahtani, Addassi, Hoteit, & Oelkers (2025). Rapid CO₂ mineralization by zeolite via cation exchange. *Scientific Reports*

Addassi, Hoteit, & Oelkers (2024). The impact of secondary silicate mineral precipitation kinetics on CO₂ mineral storage. *J. of Greenhouse Gas Control*,

Closed-Loop System



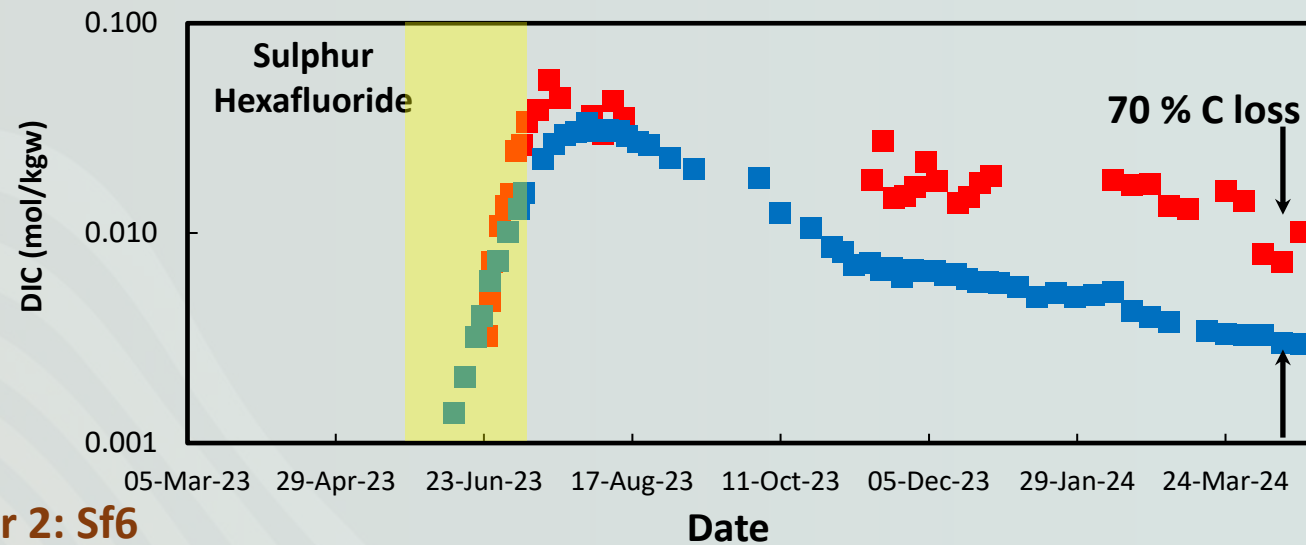
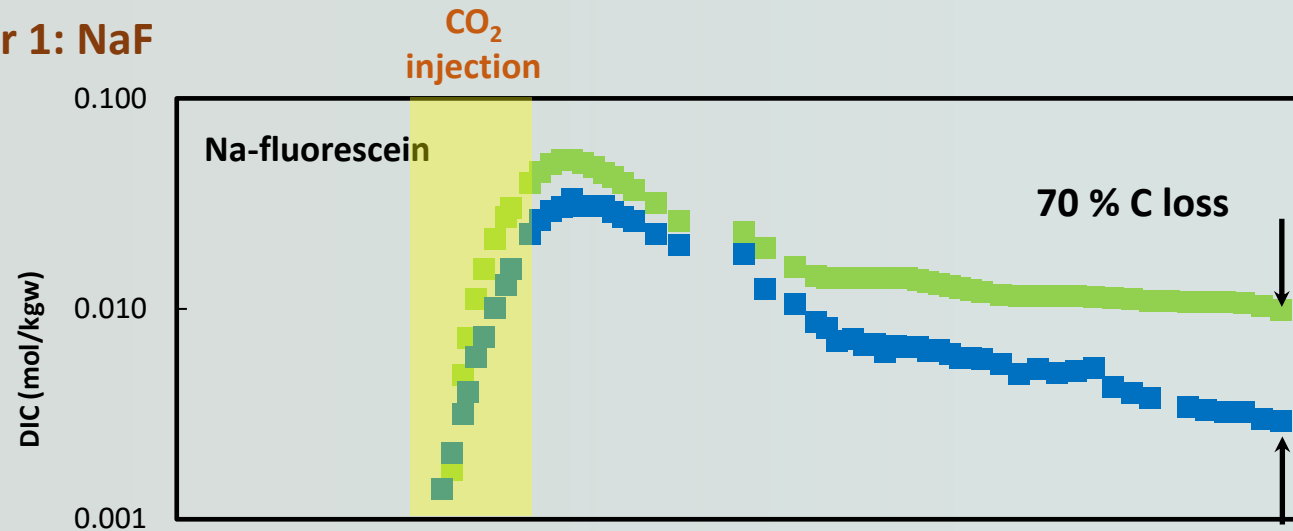
Jizan CO₂ Mineralization Pilot



Quantification of CO₂ Mineralization : 70% within 10 months



Tracer 1: NaF



Tracer 2: Sf6



Take Away

- Two-well system (130 m apart, drilled to ~1000 m)
 - Full formation (brackish) water recirculation, no external water required: critical for water-scarce arid regions)
 - Location adjacent to major point-source carbon emitters in Jizan, KSA.
- Fractured basalts (Jizan Group, 20-30Ma) at ~ 45°C, target depth >350 m, ~hydrostatic pressure conditions.
 - Pre-injection of tracer (sodium fluorescein) revealed a dual-permeability flow system consisting of few large fractures + extensive smaller fractures (matrix)
 - Effective pore volume (PV) estimated at 24,000–43,000 m³
 - 10% PV in fast flow in fracture (50–65-day residence time) + 90% PV in low-permeability matrix (255–445 day residence time)
- CO₂ dissolved directly into recycled formation water within the wellbore at 150 m depth (12–14 bar delivery pressure)
 - 131 tonnes of CO₂ injected over ~37 days at a CO₂:water-mass-ratio of 1:65 (not optimized, below solubility at reservoir conditions).
 - SF₆ co-injected with CO₂ as a conservative tracer.
- About **70%** of injected CO₂ mineralized within ~10 months
 - Dominant phases: calcite, ankerite, and siderite

Note: Carbon Mineralization is not a replacement for conventional CCS, it is strictly complementary; geological setting /conditions dictates applicability



Collaborators on Jizan Pilot:

Eric H. Oelkers, Serguey Arkadakskiy, Zeyad Ahmed, Noushad Kunnummal, Jakub Fedorik, Massimo Marchesi, Mouadh Addassi, Abdirizak Omar, Niccolo Menegoni, Sigurdur R. Gislason, Grimur Bjornsson, Davide Berno, Thomas Finkbeiner, Abdulkader Afifi, Hussein Hoteit

CO₂ subsurface mineral storage by its co-injection with recirculating water.
Nature **651**, 954–958 (2026). <https://doi.org/10.1038/s41586-026-10130-5>



Pilot site: Part of the Team